

Published on Web 06/01/2007

Direct Palladium-Catalyzed Alkynylation of N-Fused Heterocycles

Ilya V. Seregin, Victoria Ryabova, and Vladimir Gevorgyan*

Department of Chemistry, University of Illinois at Chicago, 845 West Taylor Street, Chicago, Illinois 60607-7061

Received April 18, 2007; E-mail: vlad@uic.edu

Direct transition metal-catalyzed functionalization of C-H bonds in heterocycles serves as a powerful tool for organic synthesis. This approach offers the possibility for catalytic transformation of unreactive C-H bonds into diverse functionalities, as opposed to the traditional cross-coupling methods, employing heterocyclic halides or organometallic derivatives. In particular, direct arylation and vinylation of heterocycles has already gained widespread acceptance within the synthetic community, because of its capacity to utilize simpler and cheaper precursors for the construction of complex frameworks.¹ In the last two decades, this area was rapidly growing and new types of transition metal-catalyzed direct intraand intermolecular reactions of electron-rich $^{1-3}$ and electrondefficient⁴ heterocycles, as well as simple arenes,⁴ have been developed (eq 1). These C-C bond forming reactions involve C-H arylation, heteroarylation, and vinylation. Although the majority of these methods are based on the employment of palladium catalysis in both Pd(0/II)²⁻⁴ and Pd(II/IV)⁵ modes, methods involving rhodium,⁶ platinum,⁷ and gold⁸ complexes have also been reported. Despite the vast structural complexity of the products that existing methods for direct C-H functionalization of heterocycles offer (eq 1), they are still limited to sp²-sp² carbon-carbon bondforming reactions. Herein, we report the direct palladium-catalyzed sp²-sp carbon-carbon bond-forming reaction of electron-rich heterocycles with alkynyl halides. This conceptually new approach provides straightforward and efficient access to diverse alkynyl heterocycles (eq 2).

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

$$(1)$$

It is well-established that species *i* serve as key electrophilic intermediates in Pd-catalyzed arylation/vinylation of electron-rich heterocycles.² It is also known that the similar intermediate *v*, which forms upon oxidative addition of palladium into alkynylhalides, effectively serves as an electrophilic component in Stille⁹ and Suzuki¹⁰ cross-coupling reactions with the corresponding stannyland boryl- heterocycles. We hypothesized that, analogously to *i*, electrophilic species *v* may also undergo reaction with nonfunctionalized electron-rich heterocycles.¹¹



To test the above hypothesis, we examined a number of electronrich N-fused heterocycles 1 in reaction with alkynyl halides 2 (eq 3). After certain optimization work, we found that 1, indeed, in the presence of 3 mol % of $PdCl_2(PPh_3)_2$ and 2 equiv. of KOAc in

(eq 3)				
#	R	Product		Yield, % ^a
1	Ph	Ph	3a	51 (59)
2	Si(CH ₃) ₃	Si(CH ₃) ₃	3b	62 (71)
3	Ph	H ₃ CO ₂ C Ph	3c	76 (97) ^b
4	<i>n</i> Bu	H3CU2L	3d	64 (88)
5	Si(CH₃)₃	H ₃ CO ₂ C N Si(CH ₃) ₃	3e	90 (98)
6	Si(CH ₃) ₃	Si(CH ₃) ₃	3f	87 (98)
7	Ph	Ph	3g	73 (83)
8	<i>n</i> Bu	agen -	3h	72 (89)
9	ŧ		3i	65 (76)
10	CO₂Et	COOEt	3j	64 (75)
11	Ph	EtO ₂ C, Ph Ph Ph Ph	3k	76 (97)
12	Ph	Ph N Ph	31	51 (64)
13	₽	87 S	3m	71 (85) ^b
14	CO ₂ Et		3n	63 (74)
15	Si(CH ₃) ₃	SI(CH ₃) ₃	30	59 (71)
16	A.	S C	3р	58 (67)
17	≹ −∕⊂CO₂Et		3q	50 (56) °

Table 1. Pd-catalyzed Alkynylation of N-Fused Heterocycles

 $^{\rm a}$ Isolated yields, NMR yields are in brackets. $^{\rm b}$ Trace amount of products were detected with iodo- and chloroalkynes (see Supporting Information for details). $^{\rm c}$ Yield based on recovery of starting material. 12

toluene underwent smooth coupling reaction with bromoalkynes 2 (eq 3, Table 1). Remarkably, this direct alkynylation reaction

appeared to be quite general with respect to the electron-rich N-fused¹³ heterocyclic core.¹⁴ Thus, unsubstituted- (entries 1-2)



and ester-containing indolizines (entries 3-5), pyrrolo-isoquinoline (entries 6-10), densely substituted pyrroloxazole (entry 11), and pyrroloquinoline (entries 13-17) were smoothly alkynylated to give the corresponding alkynyl heterocycles 3 in good to very high vields. Notably, bis-pyrrolo-pyrimidine underwent double fold alkynylation with excess alkynyl bromide to furnish 31 in reasonable overall yield (entry 12). This alkynylation reaction also demonstrated a remarkable tolerance toward functional groups at the bromoalkyne 2.15 Indeed, bromoalkynes possessing alkyl, aryl, alkenyl, TMS, and ester groups, were nearly equally efficient in direct alkynylation (Table 1). It should be mentioned that, in contast to bromoalkynes, their chloro and iodo counterparts were much less efficient in alkynylation, providing only trace amounts of alkynylated indolizine 3c (entry 3) and quinoline 3m (entry 13).¹⁶

We propose that the direct Pd-catalyzed C-H alkynylation of electron-rich heterocycles operates via an electrophilic substitution pathway, analogous to that previously postulated for the palladium-(0)-catalyzed arylation of electron-rich heterocycles (Scheme 1).²

Scheme 1



The mechanism involves a nucleophilic attack of the most electronrich C-3 position of heterocycle 1 at alkynylpalladium intermediate v to form iminium intermediate 4. Deprotonation of the latter furnishes the Pd^{II} intermediate 5, which upon reductive elimination produces alkynyl heterocycle 3. The electrophilic nature of the process is supported by a minor kinetic isotope effect of 1.15 observed in alkynylation of the D-labeled indolizine 6 (eq 4).¹⁶ This KIE value is in the range of those reported by us2b and others2c,d in the Pd-catalyzed arylation of electron-rich heterocycles proceeding via an electrophilic pathway.17



In summary, we developed a mild and effective method for the direct palladium-catalyzed C-H alkynylation of electron-rich heterocycles, including indolizine, pyrroloquinoline, pyrroloisoquinoline, pyrrolooxazole, and bis-pyrrolo-pyrimidine. It was shown that a variety of functional groups at bromoalkyne, such as alkyl, alkenyl, aryl, silyl, and ester, were perfectly tolerated in this alkynylation reaction. This conceptually new method for sp²-sp carbon-carbon bond-formation in heterocycles was proposed to proceed via electrophilic substitution motif.

Acknowledgment. The support of the National Institutes of Health (Grant GM-64444) is gratefully acknowledged.

Supporting Information Available: Preparative procedures, analytical and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) For recent reviews, see: (a) Seregin, I. V.; Gevorgyan, V. Chem. Soc. Rev., published online, 2007, http://dx.doi.org/10.1039/b606984n. (b) Alberico, D.; Scott, M. E.; Lautens, M. Chem. Rev. 2007, 107, 174.
- (2) For mechanistic studies on Pd-catalyzed arylation of electron-rich heterocycles, see: (a) Okazawa, T.; Satoh, T.; Miura, M.; Nomura, M. J. Am. Chem. Soc. 2002, 124, 5286. (b) Park, C.-H.; Ryabova, V.; Seregin, I. V.; Sromek, A. W.; Gevorgyan, V. Org. Lett. 2004, 6, 1159. (c) Lane, B. S.; Brown, M. A.; Sames, D. J. Am. Chem. Soc. 2005, 127, 8050. (d) Chiong, H. A.; Daugulis, O. Org. Lett. 2007, 9, 1449.
- (3) For selected references on Pd-catalyzed arylation and vinylation of electron-rich heterocycles, see: (a) Ohta, A.; Akita, Y.; Ohkuwa, T.; Chiba, M.; Fukunaga, R.; Miyafuji, A.; Nakata, T.; Tani, N.; Aoyagi, Y. Heterocycles 1990, 31, 1951. (b) Desarbre, E.; Merour, J.-Y. Heterocycles 1995, 41, 1987. (c) Lavenot, L.; Gozzi, C.; Ilg, K.; Orlova, I.; Penalva, V.; Lemaire, M. J. Organomet. Chem. 1998, 567, 49. (d) Kondo, Y.; V.; Leinare, M. J. Organomet. Chem. 1998, 507, 49. (d) Kondo, F.;
 Komine, T.; Sakamoto, T. Org. Lett. 2000, 2, 3111. (e) Glover, B.; Harvey,
 K. A.; Liu, B.; Sharp, M. J.; Tymoschenko, M. Org. Lett. 2003, 5, 301.
 (f) Mori, A.; Sekiguchi, A.; Masui, K.; Shimada, T.; Horie, M.; Osakada,
 Kawamoto, M.; Ikeda, T. J. Am. Chem. Soc. 2003, 125, 1700. (g) Li, W.; Nelson, D. P.; Jensen, M. S.; Hoerrner, R. S.; Javadi, G. J.; Cai, D.; Larsen, R. D. Org. Lett. **2003**, 5, 4835. (h) Beccalli, E. M.; Broggini, G.; Martinelli, M.; Paladino G.; Zoni, C. Eur. J. Org. Chem. **2005**, 2091. (i) Bressy, C.; Alberico, D.; Lautens, M. J. Am. Chem. Soc. 2005, 127, 13148.
- (4) For Pd-catalyzed arylation of electron-defficient heterocycles and simple arenes, see: (a) Garcia-Cuadrado, D.; Braga, A. A. C.; Maseras, F. Echavarren, A. M. J. Am. Chem. Soc. 2006, 128, 1066. (b) Lafrance, M.; Rowley, C. N.; Woo, T. K.; Fagnou, K. J. Am. Chem. Soc. 2006, 128, 8754. (c) Campeau, L.-C.; Rousseaux, S.; Fagnou, K. J. Am. Chem. Soc. 2005, 127, 18020. (d) Leclerc, J.-P.; Fagnou, K. Angew. Chem., Int. Ed. 2006, 45, 7781.
- (5) For involvement of Pd(II/IV) catalytic cycle in direct C-H functionalizations, see for example: (a) Giri, R.; Chen, X.; Yu, J.-Q. Angew. Chem., Int. Ed. 2005, 44, 2112. (b) Zaitsev, V. G.; Daugulis, O. J. Am. Chem. Soc. 2005, 127, 4156. (c) Deprez, N. R.; Kalyani, D.; Krause, A.; Sanford, (a) Lewis, J. C.; Wu, J. Y.; Bergman, R. G.; Ellman, J. A. Angew. Chem.,
- (6)Int. Ed. 2006, 45, 1589. (b) Wang, X.; Lane, B. S.; Sames, D. J. Am. Chem. Soc. 2005, 127, 4996. (c) Yanagisawa, S.; Sudo, T.; Noyori, R.; Itami, K. J. Am. Chem. Soc. 2006, 128, 11748.
- (a) Liu, C.; Han, X.; Wang, X.; Widenhoefer, R. A. J. Am. Chem. Soc. **2004**, *126*, 3700. (b) Furstner, A.; Mamane, V.; Seidel, G.; Laurich, D. Org. Synth. 2006, 83, 103.
- (a) Hashmi, A. S. K.; Schwarz, L.; Choi, J.-H.; Frost, T. M. Angew. Chem., Int. Ed. 2000, 39, 2285. (b) Li, Z.; Shi, Z.; He, C. J. Organomet. Chem. 2005, 690, 5049. (c) Nakamura, I.; Saito, S.; Yamamoto, Y. J. Am. Chem. Soc 2000 122 2661
- (a) Palmisano, G.: Santagnostino, M. Svnlett 1993, 10, 771. (b) Holling-(d) ramano, G., Sandellosino, M. Synch 1993, 7, 463.
 (10) Chan, K.-F.; Wong, H. N. C. Eur. J. Org. Chem. 2003, 1, 82.
- (11) For the Cu-catalyzed alkynylation of N-H bonds, see: (a) Frederick, M. O.; Mulder, J. Á.; Tracey, M. R.; Hsung, R. P.; Huang, J.; Kurtz, K. C. M.; Shen, L.; Douglas, C. J. *J. Am. Chem. Soc.* **2003**, *125*, 2368. (b) Dunetz, J. R.; Danheiser, R. L. Org. Lett. **2003**, *5*, 4011.
- (12) N-fused pyrroloheterocycles are known to be among the most electronrich heterocyclic cores. See, for example: Behnisch, A.; Behnisch, P.; Eggenweiler, M.; Wallenhorst, T. Indolizine. In Houben-Weyl; Thieme: Stuttgart, Germany, 1994; Vol. E6b/1, 2a, pp 323-450.
- (13) Trial attempts with non-fused heterocycles, such as pyrroles, under these reaction conditions, afforded low yields of the corresponding alkynylation products. Further studies on the scope of this transformation are underway in our group.
- (14) For preparation of bromoalkynes, see Supporting Information.
- (15) See Supporting Information for details.
- (16) Carbon-carbon bond forming reactions proceeding via C-H activation motif usually exhibit much higher KIE. See, for example: Jones, W. D. Acc. Chem. Res. 2003, 36, 140.
- The reaction was stopped at 60% conversion to avoid thermal decomposition of the product.

JA072718L